Environmental Modeling

Mathematical Model of Freezing in a Porous Medium at Micro-Scale

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Outline of the talk

- 1. Motivation and examples
- 2. Phenomenology of phase transition in porous media
 - energy balance
 - mechanical balance
 - phase tracking
 - density difference and simplifications
 - thermodynamical description
 - modified Stefan problem
 - thin layers
- 3. Conservation laws
 - mass
 - energy
 - momentum
- 4. Computational studies
 - macrosscale
 - microscale
- 5. Undergrad challenges

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- Colorado School of Mines (T.H. Illangasekare, A.C. Trautz)
- Faculty of Civil Engineering, CTU Prague (A. Žák, M. Sobotková, M. Sněhota)

Motivation

- thaw freeze cycles in nature
- release of methane from permafrost climate changes in arctic regions and consequences
- seasonal structural changes in constructions
- fabrication of advanced materials
- phase change energy storage

Examples - part I





frost heave - patterning at soil surface in regions affected by cold winter



rails after cold season



road after cold season

Examples - part II





methane gaps exploded after breaking the permafrost cap





broken permafrost layers

Experimental Evidence





coffe-cup experiment – measurements at CTU, Civil Engineering





broken permafrost layers

Measurements





measurements at CTU, Civil Engineering



measurement of temperature

Phenomenology of Phase Transitions in Porous Media

List of phenomena and factors of phase transitions in porous media:

- Heat transfer controlling mechanism
- Variations in specific volume in solid-liquid transition
- Induced forces causing deformation and motion
- Heterogeneous spatial structure
- High-curvature surface phenomena
- Gas dissolution in liquid and solid phases

Scales



Phase Diagram

Water Phase Diagram



17 known ice phases

Clapeyron equation

$$\frac{dP}{dT} = \frac{L}{T\Delta v}$$

- \bullet P thermodynamic pressure
- T absolute pressure
- L specific latent heat
- Δv specific change in volume due to the phase change

Stefan Problem with Surface Tension

Clapeyron equation incorporated into the energy balance at microscale

$$\begin{split} C \frac{\partial T}{\partial t} &= \nabla (k \nabla T) \vee \Omega_s \text{ a } \Omega_l \\ b_c(T) \mid_{\partial \Omega} &= 0 \\ T \mid_{t=0} &= T_0 \\ k \frac{\partial T}{\partial n} \mid_s -k \frac{\partial T}{\partial n} \mid_l &= L v_{\Gamma} \\ T - T^* &= -\frac{\sigma}{\Delta s} \kappa_{\Gamma} - \alpha \frac{\sigma}{\Delta s} v_{\Gamma} \\ \Omega_s(t) \mid_{t=0} &= \Omega_{so} \end{split}$$

Dirichlet boundary condition :

$$b_c(T) \mid_{\partial\Omega} = u \mid_{\partial\Omega} - g_0$$

or Neumann boundary condition:

$$b_c(T) \mid_{\partial\Omega} = \frac{\partial u}{\partial n} \mid_{\partial\Omega} -g_1$$

 $\begin{array}{lll} \lambda_s, \lambda_l & \mbox{heat conductivity} \\ \sigma = f(T^*) & \mbox{surface tension} \\ \hline \kappa_{\Gamma} & \mbox{mean curvature of the hypersurface } \Gamma(t) \\ \mathbf{n}_{\Gamma} & \mbox{normal unit vector to } \Gamma(t) \mbox{pointing out of } \Omega_s \\ v_{\Gamma} & \mbox{normal unit vector to } \partial\Omega \mbox{ pointing out of } \Omega \\ \eta_L(t), \ \Omega_S(t) \mbox{liquid/solid subdomain} \\ T = T(t, x) & \mbox{temperature} \\ L & \mbox{latent heat} \end{array}$

 Δs (...* ...)

$$\rho_{I}c_{I}\frac{\partial u}{\partial t} = \nabla \cdot (\lambda_{I}\nabla u)$$

$$\Gamma$$

$$\Omega_{S}$$

$$\Omega_{I}$$

$$\Omega_{S}$$

$$\Omega_{I}$$

$$\rho_{S}c_{S}\frac{\partial u}{\partial t} = \nabla \cdot (\lambda_{S}\nabla u)$$

Phenomenology

Thin Liquid Layers



Layered structures of phase interface (1) = ice, (2) = grain, (3) = liquid

Liquid layer occurs when **surface tensions** of solid/liquid + liquid/grain < solid/grain

Thin Layer Understanding



Disordered upper layer of water molecules

Premelting and porous media



Schematics of thermal regelation

Force acting on solid object in the direction of temperature gradient

$$\mathbf{F}_T = \frac{\varrho_s L}{T^*} \int_{Volume} \nabla T dx$$

Rempel A. W. Microscopic and environmental controls on the spacing and thickness of segregated ice lenses. *Quaternary Research*, 75:316–324, 2011.

Thin Liquid layer

Poromechanics

Assuming linear elasticity and very small displacements

Strain in Ω_G and Ω_P

$$\epsilon_{ij}^{el} = \frac{1}{2} \left(\frac{\partial u_i}{\partial x_j} + \frac{\partial u_j}{\partial x_i} \right)$$

Stress by Hook's law in Ω_G

$$\sigma_{ij} = \frac{E_G}{1+\nu} \left(\epsilon_{ij}^{el} + \frac{\nu}{1-2\nu} \epsilon_{kk}^{el} \delta_{ij} \right)$$

Stress expression with expansivity in Ω_P

$$\sigma_{ij} = \frac{E_P(p)}{1+\nu} \left(\epsilon_{ij}^{el} + \frac{\nu}{1-2\nu} \epsilon_{kk}^{el} \delta_{ij} \right) + F_{exp}(p) \mathbb{I}$$

where

$$E_G = const.,$$
 $E_P(p) = pE_{melt} + (1-p)E_{solid},$ $F_{exp}(p) = c_{exp}p$

Newton's force balance law

$$arrho rac{\partial^2 u_i}{\partial t^2} - rac{\partial \sigma_{ij}}{\partial x_j} = 0$$

with natural boundary conditions and with continuity of displacement and derivatives across $\Omega_G \cap \Omega_P$

Poromechanics

Conservation Laws - geometry



Space domain consists of grains Ω_G and pores filed by liquid Ω_L and solidified phase Ω_S

Deformation $\mathbf{u} = \mathbf{u}(t, x)$ in all parts of the domain motivates the use of Lagrangian coordinates

Conservation Laws - general

Volume balance

$$\bar{\phi}(t,x) + \phi(t,x)\nabla \cdot \mathbf{v}(t,x) + \nabla \cdot \xi(t,x) - \Sigma(t,x) = 0 \quad \text{in } \mathcal{V} - \mathcal{S}(t)$$

Interface balance

$$[\![\phi(t,x)(\mathbf{v}(t,x)-\mathbf{u}(t,x))-\xi(t,x)]\!]\cdot\mathbf{n}=0 \quad \text{on } \mathcal{S}(t)$$

- $\dot{\bar{\phi}}$ material derivative of quantity ϕ
- $\mathbf{v}(t,x)$ velocity of material point
- $\xi(t,x)$ surfacial flux of quantity ϕ
- $\Sigma(t,x)$ volumic source density of quantity ϕ
- $\bullet~ \mathcal{V}$ investigated volume
- $\mathcal{S}(t)$ boundary moving with velocity $\mathbf{u}(t,x)$
- **n** normal vector to $\mathcal{S}(t)$

Applied to:

- mass described by mass density ϱ
- momentum described by momentum density $\rho \mathbf{v}$
- $\bullet~$ energy described by internal energy density and characterized by absolute temperature T

Conservation Laws - constitutive relations

Stress in grain matrix

$$\sigma_{ij} = \frac{E_G}{1+\nu} \left(\epsilon_{ij}^{el} + \frac{\nu}{1-2\nu} \epsilon_{kk}^{el} \delta_{ij} \right), \quad \epsilon_{ij}^{el}(t,x) = \frac{1}{2} \left(\frac{\partial u_i}{\partial x_j} + \frac{\partial u_j}{\partial x_i} \right)$$

Stress in solid phase

$$\sigma_{ij} = \frac{E_S}{1+\nu} \left(\epsilon_{ij}^{el} + \frac{\nu}{1-2\nu} \epsilon_{kk}^{el} \delta_{ij} \right) + \beta \delta_{ij}$$

where β is the volume expansion factor

Stress in Newtonian incompressible liquid phase

$$\sigma_{ij} = -p(t, x)\delta_{ij} + 2\mu \mathbf{d}_{ij}(t, x), \quad \mathbf{d}_{ij}(t, x) = \frac{1}{2} \left(\frac{\partial v_i}{\partial x_j} + \frac{\partial v_j}{\partial x_i} \right)$$

Fourier law for heat flux and expression for internal energy of solid

$$\xi_{heat} = -k\nabla T, \qquad u(t,x) = \int_{T_0}^T C(T) \mathrm{d}T$$

where k is heat conduction coefficient, c is specific heat capacity

Enthalpy of liquid including latent heat L

$$h(t,x) = u(t,x) + L$$

Conservation laws

Macro-Scale Model

Designed for large macroscale, based on empirical melting point depression by pores with high-curvature boundaries

Heat conduction with empirical latent heat source term

$$C\frac{\partial T}{\partial t} + L\frac{\partial \theta(T)}{\partial t} = \nabla \cdot (k\nabla T), \quad \theta(T) = \begin{cases} \eta & \text{for } T \ge T^D \\ \eta \left| \frac{T^D - 273.15}{T - 273.15} \right|^b & \text{for } T < T^D \end{cases}$$

where - η is porosity, $T^D < 273.15$ melting point depression, b>0 empirical constant

Momentum balance at large scale (homogenized)

$$arrho rac{\partial^2 u_i}{\partial t^2} - rac{\partial \sigma_{ij}}{\partial x_j} = 0$$

$$\sigma_{ij} = \frac{E}{1+\nu} \left(\epsilon_{ij}^{el} + \frac{\nu}{1-2\nu} \epsilon_{kk}^{el} \delta_{ij} \right) + \beta(T) \delta_{ij}, \quad \beta = \begin{cases} \bar{\beta} & \text{for } T \leq T^D \\ 0 & \text{for } T > T^D \end{cases}$$

Reference and with references therein:

Žák A., Beneš M. and Illangasekare T. H. Analysis of Model of Soil Freezing and Thawing. *IAENG International Journal of Applied Mathematics*, Volume 43, Issue 3, pp. 127–134, September 2013

Macro-Scale Computational Studies - building heave



Strain evolution of the concrete construction during ground soil freezing

An illustrative example explaining purpose of modeling

Macro-Scale Computational Studies - vertical soil behavior



Vertical heterogeneous soil structure filled by liquid and solid phase with indication of temperature at a particular time moment. Cooling set from above. Domain deformation visually magnified.

Symmetric Micro-Scale Model - results 1



Temperature distribution in a symmetric pore domain under vertical external temperature gradient

Symmetric Micro-Scale Model - results 2



Solidification in a triangular symmetric pore domain under vertical external temperature gradient

Phase-Field Micro-Scale Model

Derived for representative microscale on a general pore domain.

- a new function p = p(t, x) with values in $\langle 0, 1 \rangle$
- solid phase $\Omega_S(t) \equiv p(t, x) = 1$
- liquid phase $\Omega_L(t) \equiv p(t, x) = 0$
- phase interface $\Gamma(t) \equiv p(t, x) = \frac{1}{2}$
- function p is obtained from the Allen-Cahn equation

Reference:

Žák A. and Beneš M. Micro-Scale Model of Thermomechanics in Solidifying Saturated Porous Media, *Acta Physica Polonica* 134 (3), 2018, pp. 678–682.

Computational studies - small pack



Color scale indicates distribution of temperature at selected time moments t = 70, t = 90 and t = 100

Computational Studies - part 1



Formation of solid phase, coalescence of two nucleated sites, growth of common pattern and its positioning at pore center

Nucleation sites defined by the initial condition

Computational Studies - part 2



Pore filled by liquid (in blue) and solid (in yellow), structural changes oversized and indicated in green Formation of solid phase in complex pore geometry

Nucleation site defined by the initial condition

Challenges - for undergrads I



Gas release from porous media - multiphase flow described by the Darcy law.

 $\mathbf{j}_{lpha} = -\mathbb{K}_{lpha}
abla p_{lpha}, \quad ext{for the phase } lpha$

Challenges - for undergrads II



Modeling the **frost heave** during soil freezing.

Challenges - for undergrads III



Transport along surfaces during soil freezing.

Question

Henry's law for maximum gas concentration solvable in a liquid

$$C = H \exp(Q(\frac{1}{T} - \frac{1}{T_{ref}}))Mp$$

- $C \ kg/m^3$ gas concentration
- $p \ Pa$ gas pressure
- $T \ K$ system temperature
- $H \ mol/m^3/Pa$ Henry's constant
- $M \ kg/mol$ molar gas mass
- Q K constant

For argon at p = 3atm = 300kPa is maximum concentration $0.147kg/m^3$, i.e. the experimental container of 28ml has maximum of 4.129mg of argon.